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\u0026 Percent Yield - Chemistry

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OBJECTIVES:

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- Explain how balanced equations apply to both chemistry and everyday life.

3.

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describes how much of each

reactant is required to make a

product. Stoichiometry tells us if

chemical processes are efficient .

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2. STOICHIOMETRY Nearly
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manufactured from chemicals -

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quantitative relationships between the amounts of reactants used and products formed by a chemical reaction. It is based on the law of conservation of mass, which says that matter is not created or destroyed. Chapter_12_Stoichio

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