

Chromatography

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Chromatography is a laboratory technique for the separation of a mixture. The mixture is dissolved in a fluid (gas, solvent, water,...) called the mobile phase, which carries it through a system (a column, a capillary tube, a plate, or a sheet) on which is fixed a material called the stationary phase.

Chromatography - Wikipedia

Chromatography, technique for separating the components, or solutes, of a mixture on the basis of the relative amounts of each solute distributed between a moving fluid stream, called the mobile phase, and a contiguous stationary phase. The mobile phase may be either a liquid or a gas, while the stationary phase is either a solid or a liquid.

chromatography | Definition, Types, & Facts | Britannica

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Chromatography is the most widely used separation technique in chemical laboratories, where it is used in analysis, isolation and purification, and it is commonly used in the chemical process industry as a component of small and large-scale production.

Chromatography - an overview | ScienceDirect Topics

The term chromatography literally means color writing, and denotes a method by which the substance to be analyzed is poured into a vertical glass tube containing an adsorbent, the various components of the substance moving through the adsorbent at different rates of speed, according to their degree of attraction to it, and producing bands of color at different levels of the adsorption column.

Chromatography | definition of chromatography by Medical ...

Chromatography is an analytical technique used to separate mixture of chemical substances into its individual compounds. Different types of chromatography are used in lab. e.g. column chromatography, thin-layer chromatography, gas chromatography etc. Chromatography consists of two phases: one mobile phase and one contiguous stationary phase.

Chromatography Definition, Types And Examples Of ...

‘Chromatography’ is an analytical technique commonly used for separating a mixture of chemical substances into its individual components, so that the individual components can be thoroughly analyzed.

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Principles of chromatography | Stationary phase (article ...

Chromatography not only separates out your target molecule, it concentrates it and offers a variety of downstream applications as a result. Learn more about the different separation techniques and the chromatography solutions that Thermo Fisher Scientific offers.

Chromatography | Thermo Fisher Scientific - US

Chromatography is actually a way of separating out a mixture of chemicals, which are in gas or liquid form, by letting them creep slowly past another substance, which is typically a liquid or solid.

How does chromatography work? - Explain that Stuff

Chromatography involves taking some kind of mixture and using either solid or liquid to separate it out into its different parts. There are many different kinds of chromatography, but they all rely on having a mobile phase and a stationary phase. Let's go over how paper chromatography works, since this is the simplest kind.

Basics of chromatography (video) | Khan Academy

Paper chromatography, in analytical chemistry, technique for separating dissolved chemical substances by taking advantage of their different rates of migration across sheets of paper. It is an inexpensive but powerful analytical tool that requires very small quantities of material.

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