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Subtraction pages 51-54

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WORKSHEET Using the Periodic Table

1. Identify the number of valence electrons in each of the following elements. a. Ne d. Mg f. Cl h. Si
2. Identify the energy level of the valence electrons in each of the following elements. a. c. d. e. H Ar 1 3.

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WORKSHEET Determining Numbers of

Ions Use with Chapter 7, Section 7.3 1.

Write a simple word equation that illustrates what must be true of total positive charge and total negative

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Answers
charge in an ionic compound. 2. How many potassium ions (group 1) would be related to balance the charge of each of the following in a compound?

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WORKSHEET 1 MATH SKILLS

TRANSPARENCY WORKSHEET 2

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with Chapter 2, Visualizing the Use
with Chapter 3, Section 2.4
Conservation of Mass Section 3.2 1.

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The exponents are added. $(2.00 \times 10^4) = \times 10^{3+4} = 8.00 \times 10^7$ Division

When numbers in scientific notation are divided, only the number is divided. The exponents are subtracted. $9.60 \times 10^7 \div 1.60 \times 10^4$

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Masters $9.60 \times 10^7 \div 1.60 = 6.00 \times 10^3$

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Chemistry: Matter and Change Name

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TRANSPARENCY WORKSHEET

Operations with Scientific Notation I.

Perform the following operations and express the answers in scientific notation.

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WORKSHEET Use with Chapter 1,

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Answers Section 11.1 MATH SKILLS

TRANSPARENCY WORKSHEET 16

Section 11.2 Determine the mole ratios for each of the following balanced chemical equations.

$1. 2C(s) + O_2(g) \rightarrow 2CO(g)$ 1 mol O_2 2 mol C
2 mol CO / 2 mol C 2 mol C / 1 mol O_2
2 mol CO / 1 mol O_2 2 mol C / 2

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Answers
mol CO 1 mol O 2 /2 mol CO 2 2. WO 3
(s) + 3H 2 (g)0W(s) 2

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MASTER 16

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