

## Mole Ratio Pogil Answer Key

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Pogil Chemistry Answer Keys Mole Both answers should be about  $6.02 \times 10^{23}$ . The quantity,  $6.02 \times 10^{23}$  is Avogadro's constant and we call it the 'mole'.

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1 mole =  $6.022 \times 10^{23}$  particles or 602 200 000 000 000 000 000 000 particles (More than you could count in a lifetime!) This number is called Avogadro's number, named after Amedeo Avogadro.

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